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Isoprenylated Phenolics from Roots of Artocarpus heterophyllus

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Thirteen compounds were isolated from *Artocarpus heterophyllus*, including one new di-isoprenylated stilbene, (\pm)-artoheterin A (**1**), and twelve known isoprenylated phenolics, artocarbene (**2**), 3',5',2,4-tetrahydroxy-4'-(3-methyl-2-butenyl) stilbene (**3**), cudrastilbene (**4**), 3-(3-methyl-2-butenyl) luteolin (**5**), 5,7,2',4'-tetrahydroxy-6-(3-methyl-2-butenyl) flavone (**6**), 6-(3-methyl-2-butenyl) apigenin (**7**), artochamin B (**8**), 5'-hydroxycudraflavone A (**9**), artonins A and B (**10-11**), heterophyllin (**12**), and 8-(3-methyl-2-butenyl)-6,7-dihydroxycoumarin (**13**). The structures of these compounds were identified by spectroscopic methods. All isolated compounds were screened for their inhibitory abilities against cathepsin K (CatK). Of them, compounds **1-4** and **8-13** were found to have suppression capabilities against CatK with IC₅₀ values ranging from 1.9 to 73.7 μ M.

Keywords: Moraceae, Artocarpus heterophyllus, Cathepsin K inhibitory activity, Artoheterin A, Isoprenylated phenolics, Stilbene.

Artocarpus heterophyllus (Moraceae), an evergreen arbor tree, is mainly distributed in tropical and subtropical regions of Asia, including Southern China. In China, A. heterophyllus is popularly cultivated for its edible fruits, while its root has been used as a folk medicine for subduing swelling and detoxicating [1]. Previous phytochemical studies on A. heterophyllus have displayed the occurrence of a variety of flavonoids and 2-arylbenzofurons with cytotoxicity [2], tyrosinase inhibitory [3], anti-inflammatory [4], antioxidant [5], and anti-respiratory burst activities [6]. Cathepsin K (CatK) is the most important enzyme for osteoclast-mediated bone resorption and inhibition of CatK is generally regarded as an effective approach for the treatment of osteoporosis [7]. Recently, we reported a variety of isoprenylated flavonoids with CatK inhibitory activities from A. styracifolius [8]. As a part of our continuing research on Artocarpus plants, the investigation on chemical constituents with CatK inhibitory activities from A. heterophyllus were performed. Herein, we described the isolation, structural elucidation, and CatK inhibitory capabilities of one new di-isoprenylated stilbene, artoheterin A (1), as well as twelve known isoprenylated phenolics, artocarbene (2), 3',5',2,4tetrahydroxy-4'-(3-methyl-2-butenyl) stilbene (3), cudrastilbene (4), 3-(3-methyl-2-butenyl) luteolin (5), 5,7,2',4'-tetrahydroxy-6-(3methyl-2-butenyl) flavone (6), 6-(3-methyl-2-butenyl) apigenin (7), artochamin B (8), 5'-hydroxycudraflavone A (9), artonins A and B (10-11), heterophyllin (12), and 8-(3-methyl-2-butenyl)-6,7dihydroxycoumarin (13) (Figure 1).

(±)-Artoheterin A (1), a brownish red amorphous powder, has a positive reaction with ferric chloride reagent, proving the presence of the phenolic moiety. Its molecular formula was determined to be $C_{26}H_{32}O_4$ by HR-ESI-MS at m/z 407.2220 ([M-H]⁻, calcd for $C_{26}H_{31}O_4$, 407.2228). The IR spectrum of 1 showed absorptions for hydroxyl group (3404 cm⁻¹), aliphatic chain (2930 and 2857 cm⁻¹), and aromatic ring (1601, 1455 and 1377 cm⁻¹). The ¹H NMR spectrum (Table 1) of 1 exhibited signals for an aromatic ABX spin



Figure 1: Structures of compounds 1-13. An asterisk designates relative configuration for the specific chiral carbon.

system (ring A) at δ 7.25 (1H, d, J = 8.0 Hz, H-6), 6.38 (1H, br. s, H-3), and 6.39 (1H, overlap, H-5), a 3-methyl-2-butenyl (isoprenyl) group at δ 4.94 (1H, t, J = 6.1 Hz, H-21), 3.05 (1H, m, H_a-20), 3.03 (1H, m, H_b-20), 1.50 (3H, s, H-24), and 1.44 (3H, s, H-23), an aromatic proton (ring B) at δ 6.45 (1H, s, H-12), four tertiary methyl groups (two oxygenated) at δ 3.82 (3H, s, MeO-13), 3.78 (3H, s, MeO-11), 1.07 (3H, s, H₃-18), and 0.74 (3H, s, H₃-19),

Table 1: ¹H and ¹³C NMR data of compound 1 and artochamin I [9] in acetone-d₆.

Position	1 ^a		artochamin I ^b		
	$\delta_{\rm H}({\rm m},J{\rm in}{\rm Hz})$	$\delta_{ m c}$	$\delta_{\rm H}({\rm m},J{\rm in}{\rm Hz})$	$\delta_{ m C}$	
1		120.2		134.2	
2		158.6	6.75 (d, 1.8)	116.3	
3	6.38 (br. s)	103.0		145.8	
4		157.2		144.4	
5	6.39 (overlap)	106.9	6.76 (d, 8.0)	115.8	
6	7.25 (d, 8.0)	129.2	6.58 (dd, 1.8, 8.0)	120.2	
7	3.31 (d, 5.8)	51.4	2.77 (d, 5.7)	58.9	
8	4.07 (br. t, 6.4)	44.0	3.94 (t, 5.7)	45.9	
9		150.5		150.2	
10		118.0		118.4	
11		161.3		158.9	
12	6.45 (s)	94.8	6.45 (s)	95.4	
13		155.4		155.6	
14		124.7		125.0	
15	2.90 (overlap)	30.8	3.00 (dd, 1.8, 17.0)	30.9	
	2.86 (overlap)		2.84 (dd, 9.7, 17.0)		
16	2.70 (m)	45.1	2.71 (m)	45.6	
17		39.1		39.3	
18	1.07 (s)	26.1	0.97 (s)	26.4	
19	0.74 (s)	27.6	0.71 (s)	27.7	
20	3.05 (m)	26.6	2.97 (dd, overlap)	26.8	
	3.02 (m)		2.86 (dd, 5.7, 14.5)		
21	4.94 (t, 6.1)	124.6	4.88 (br. t, 5.7)	124.7	
22		130.4		131.0	
23	1.44 (s)	17.8	1.39 (br. s)	18.0	
24	1.50 (s)	25.8	1.48 (br. s)	26.0	
НО-2	7.99 (br. s)		7.71 (br. s)*		
HO-4	7.99 (br. s)		7.68 (br. s)*		
MeO-11	3.78 (s)	56.3	3.76 (s)	56.7	
MeO-13	3.82 (s)	55.5	3.81 (s)	55.9	

^a Bruker Avance 600 spectrometer; chemical shifts (ppm) referred to (CD₃)₂CO ($\delta_{\rm H}$ 2.04; $\delta_{\rm C}$ 206.0).

^b Bruker DRX 500 spectrometer; chemical shifts (ppm) referred to (CD₃)₂CO ($\delta_{\rm H}$ 2.04; $\delta_{\rm C}$ 206.0).

* The assignment may be interchanged.



Figure 2: a) Key HMBC (H \rightarrow C) correlations of 1; b) Key NOESY (H \leftrightarrow H) correlations of 1.

a methylene groups at δ 2.90 (1H, overlap, H_a-15) and 2.86 (1H, overlap, H_b-15), as well as three aliphatic methines at δ 3.31 (1H, d, J = 5.8 Hz, H-7), 4.07 (1H, br. t, J = 6.4 Hz, H-8), and 2.70 (1H, m, H-16). These ¹H NMR data was highly similar to those of the known compound, artochamin I [9]. A comparison of the ¹H NMR data (Table 1) of **1** with artochamin I showed that the main difference is the chemical shift dislocations of the three protons from ring A, indicating they share the same ring B, C, and D, but

had different substitution on ring A. This was supported by the ¹³C NMR data of **1** also similar to that of artochamin I, except for the carbons from ring A and C-7 (Table 1). The interpretation of the DEPT, HSQC, and HMBC spectra of **1** allow us to determine the substitution pattern of ring A as 2',4'-dihydroxy substitution (Table 1, Figure 2). For the relative configuration of **1**, the NOESY cross peaks of H-8 with H-16, H_a-15, and H₃-19 and H-7 with H₃-18 and H_b-15 indicated a *cis* arrangement between H-8 and H-16. In addition, the optical rotation value of **1** was zero, suggesting **1** was obtained as a racemate. The existence of enantiomers was further supported by a follow-up chiral HPLC analysis, which displayed a pair of chromatographic peak with an intensity ratio of ca. 1:1. Thus, the structure of **1** was established as shown (Figure 1) and **1** was named (±)-artoheterin A.

By comparison of their ¹H, ¹³C NMR and ESI-MS spectral data with those reported in literatures, the other twelve known isoprenylated compounds (2-13) were identified as artocarbene (2) [10], 3',5',2,4-tetrahydroxy-4'-(3-methyl-2-butenyl) stilbene (3) [11], cudrastilbene (4) [12], 3-(3-methyl-2-butenyl) luteolin (5) [13], 5,7,2',4'-tetrahydroxy-6-(3-methyl-2-butenyl) flavone (6) [14], 6-(3-methyl-2-butenyl) apigenin (7) [15], artochamin B (8) [16], 5'-hydroxycudraflavone A (9) [17], artonins A and B (10-11) [18], heterophyllin (12) [19], and 8-(3-methyl-2-butenyl)-6,7-dihydroxycoumarin (13) [20].

The inhibitory effects of all isolated compounds on CatK were evaluated by a fluorescence colorimetric assay described previously [8]. The results are summarized in Table 2. Compounds 1-4, and 8-13 showed inhibitory capacities to CatK with the IC₅₀ values ranging from 1.9 to 73.7 μ M. The other compounds showed no activities up to a highest concentration of 100 μ M.

Table 2: Inhibitory activities of 1-13 on cathepsin K.

Compounds	IC50 (µM)	Compounds	IC50 (µM)	
1	73.7	8	1.9	
2	9.6	9	2.0	
3	10.9	10	1.9	
4	24.6	11	9.0	
5	ND	12	48.4	
6	ND	13	13.2	
7	ND	$E64^{b}$	0.003	

a ND, not detectable. b Positive control.

Experimental

General: IR spectrum was executed on Shimadzu Iraffinity-1 FTIR spectrometer with KBr disc (Shimadzu Co., Kyoto, Japan). Optical rotation was determined on a JASCO P-1020 polarimeter (JASCO International Corp., Ltd, Tokyo, Japan) at room temperature. NMR spectra were recorded on a Bruker Avance 600 spectrometer (Bruker Biospin, Rheinstetten, Germany) using acetone- d_6 and/or acetone- d_6 + D₂O as solvent. HR-ESI-MS analyses were implemented on an AB SCIEX Triple TOF^{TM} 5600+ mass spectrometer (AB SCIEX Co., Framingham, MA USA). Column chromatography (CC) was performed on HP-20 (75-150 µm, Mitsubishi Chemical Co., Tokyo, Japan), ODS gel (75-150 µm, YMC Co., Kyoto, Japan), MCI GEL CHP20P (75-150 µm, Mitsubishi Chemical Co., Tokyo, Japan) and Sephadex LH-20 (25-100 µm, GE Healthcare Bio-Sciences, Amersham, Sweden). Precoated TLC plates with silica gel GF254 (10-40 µm; Yantai Jiang You silicone Development Co., Ltd., Yantai, China) were used to detect the purity of the isolates achieved by coating with 10% H₂SO₄ in EtOH, followed by heating. Preparative HPLC was executed on a LC3000 liquid chromatograph (Beijing Tong Heng Innovation Technology Co., Ltd, Beijing, China) armed with an ODS column (5 µm, 250 mm × 30 mm i.d., Sepax Technologies, Inc.). Chiral HPLC analysis was performed on a Phenomenex Lux Cellulose-2 column (5 μ M, 4.6 \times 250 mm) using acetonitrile-H₂O (4:1, v/v), at 1.2 mL/min.

Plant material: The roots of *A. heterophyllus* were collected from Nanning city, Guangxi Zhuang Autonomous Region, China, in March 2013 and identified by Lv Shihong, associate researcher of Guangxi Institute of Botany, Chinese Academy of Sciences. The voucher specimen (TCM20130101) was deposited in the Herbarium of the Department of Pharmacognosy, Research Center of Natural Resources of Chinese Medicinal Materials and Ethnic Medicine, Jiangxi University of Traditional Chinese Medicine.

Extraction and isolation: The air-dried and powdered roots of A. heterophyllus (17.0 kg) were extracted with 95% EtOH three times (140 L for each extraction) at room temperature. The filtrate was evaporated in vacuo to produce a residue (1.5 kg), which was suspended in H₂O, and then partitioned successively with petroleum ether (PE), CHCl₃, EtOAc, and n-BuOH to obtain the PE-soluble, CHCl₃-soluble, EtOAc-soluble and *n*-BuOH-soluble portions, respectively. The CHCl₃-soluble (574.0 g) was fractionated by HP-20 macroporous resin column chromatography (CC) eluted successively with a gradient of EtOH/H₂O $(0\%\rightarrow 100\%)$ to give eleven fractions (Fr. H1-H11). Fr. H6 (57.3 g) was subjected to CC over ODS eluted by EtOH/H₂O ($40\% \rightarrow 100\%$) to yield six subfractions (Fr. H6O1-H6O6). Fr. H6O3 (13.7 g) was separated by CC over MCI CHP-20P resin eluted with EtOH/H2O $(50\% \rightarrow 100\%)$ to obtain seven subfractions Fr. H6O3M1-H6O3M7. Fr. H6O3M2 (1.3 g) was further fractioned by CC on Sephadex LH-20 gel eluted with MeOH to give five subfractions Fr. H6O3M2L1-H6O3M2L5. Fr. H6O3M4 (1.5 g) was further fractioned by CC on Sephadex LH-20 gel eluted with MeOH to give seven subfractions Fr. H6O3M4L1-H6O3M4L7. Fr. H6O4 (2.9 g) was separated by CC over MCI CHP-20P resin eluted with EtOH/H₂O (70% \rightarrow 100%) to obtain three subfractions Fr. H6O4M1-H6O4M3. Fr. H6O4M3 (674.0 mg) was further fractioned by CC on Sephadex LH-20 gel eluted with MeOH to give three subfractions Fr. H6O4M3L1-H6O4M3L3. Fr. H5 (10.0 g) was fractioned by CC on Sephadex LH-20 gel eluted with MeOH to yield ten subfractions (Fr. H5L1-H5L10). Fr. H5L5 (1.5 g) was subjected to CC over ODS eluted by MeOH/H2O $(40\% \rightarrow 100\%)$ to obtain three subfractions Fr. H5L5O1-H5L5O3. Fr. H5L6 (1.2 g) was separated by CC over MCI CHP-20P resin eluted with MeOH/H₂O ($40\% \rightarrow 100\%$) to obtain two subfractions Fr. H5L6M1-H5L6M2. Fr. H6O3M2L2 (148.0 mg) was further purification by preparative HPLC eluting with acetonitrile-H2O (4:6, v/v) to yield 5 (6.0 mg, t_R 33 min). In the same way, 4 (7.1 mg, MeOH-H₂O (3:7, v/v), t_R 45 min) and 3 (284.5 mg, MeOH- H_2O (2:3, v/v), t_R 55 min), 6 (11.0 mg, acetonitrile- H_2O (4:6, v/v), $t_{\rm R}$ 58 min), 7 (2.8 mg, acetonitrile-H₂O (1:1, v/v), $t_{\rm R}$ 30 min), 8 (15.6 mg, acetonitrile-H₂O (6:4, v/v), t_R 30 min), 9 (15.1 mg, acetonitrile-H₂O (5:5, v/v), t_R 38 min), 11 (22.0 mg, acetonitrile- H_2O (7:3, v/v), t_R 25 min), 12 (3.0 mg, acetonitrile- H_2O (3:2, v/v), $t_{\rm R}$ 20 min), 10 (3.0 mg, acetonitrile-H₂O (4:1, v/v), $t_{\rm R}$ 30 min), and 1 (2.0 mg, acetonitrile-H₂O (4:1, v/v), t_R 20 min) were obtained from subfractions H5L5O3, H5L6M2, H6O3M2L3, H6O3M4L4, H6O3M4L7, H6O3M7, H6O3M6, H6O4M3L2, H6O4M3L4 and H6O6, respectively. 13 (4.0 mg) and 2 (6.9 mg) were obtained by preparative thin layer chromatography developed with CHCl₃-MeOH (10:1, v/v) from H5L3 and H5L6M2, respectively.

Evaluation for inhibitory capability against CatK: The inhibitory abilities of compounds 1-13 against CatK were performed on a fluorometer microplate reader based on the specific reaction of CatK with its substrate Z-GPR-AMC, as described previously [8], with some modifications. In brief, the compounds were diluted with the buffer (pH 6.8, 400 mM NaH₂PO₄-Na₂HPO₄ buffer, 8 mM Dithiothreitol, and 4 mM EDTA), and mixed with CatK (diluted in 0.1% Brij35) for 5-10 min at 25°C. Then, substrate Z-GPR-AMC was added to start the reaction. The final concentration of substrate Z-GPR-AMC was 20 µM in the assay mixture, Cat K was 14 nM. After incubation for 120 min at 37°C, the produced fluorescence was monitored on a SPECTROstar Omega fluorescent plate reader, emission at 460 nm after excitation at 355 nm. E64 (N-[N-(L-3-Trans-carboxirane-2-carbonyl)-L-leucyl]-agmatine), a general cysteine protease inhibitor, was used as the positive control to ensure the validity of the in vitro screening system.

Artoheterin A (1)

Brownish red amorphous powder. IR (KBr): 3404, 2930, 2857, 1601, 1455, 1377, 1319, 1204, 1121, 1018. UV λ_{max} (MeOH) nm (log ε): 284 (4.01), 331(3.75) ¹H and ¹³C NMR: Table 1.

HR-ESI-MS: m/z [M-H]⁻ calcd for C₂₆H₃₁O₄: 407.2228; found: 407.2220 and m/z [M+H]⁺ calcd for C₂₆H₃₃O₄: 409.2373; found: 409.2375.

Supplementary data: ¹H and ¹³C NMR, DEPT 135, HSQC, HMBC, UV, IR, and HR-ESI-MS (positive and negative ion modes) of 1.

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